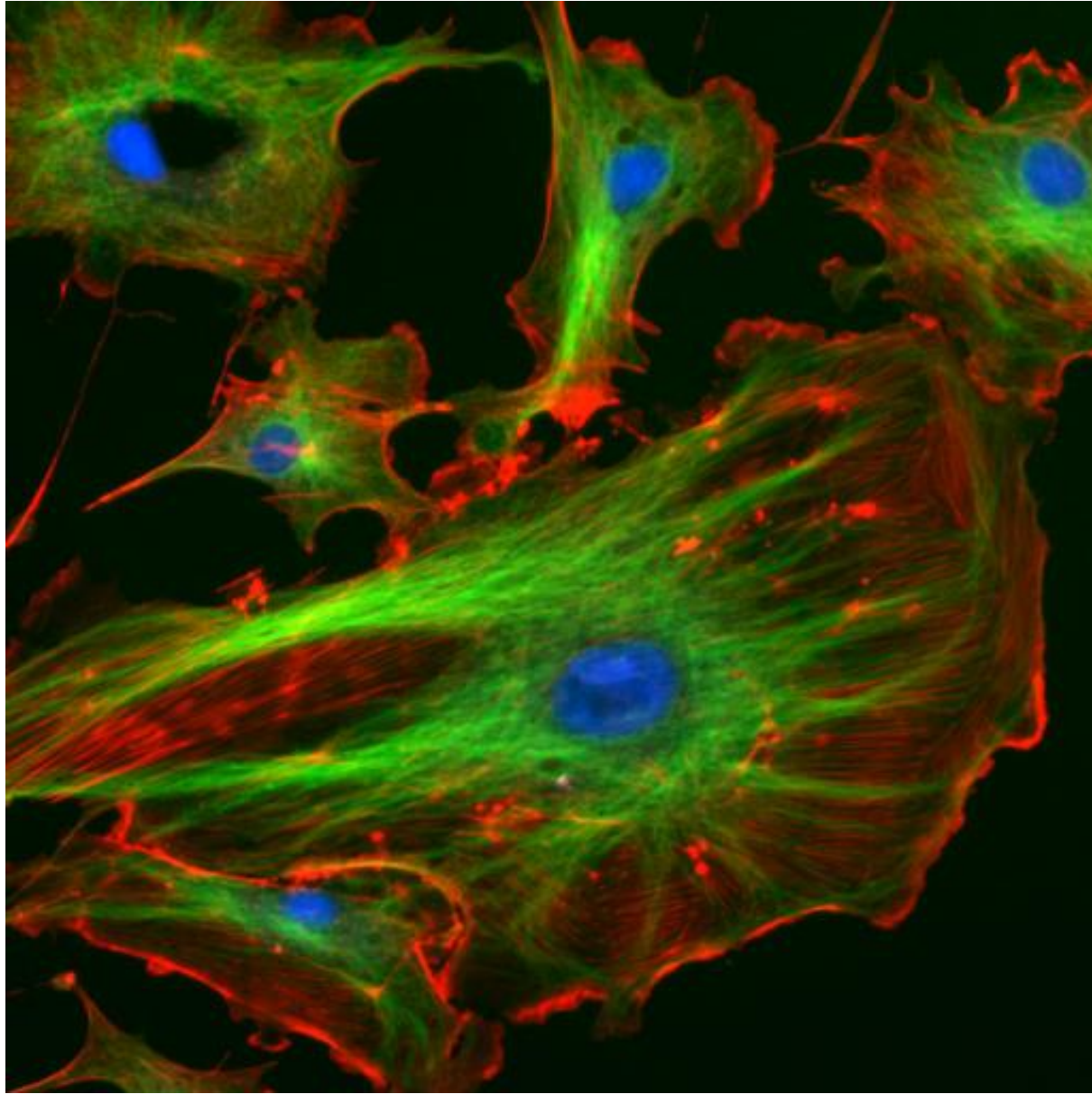


FLUORESCENCE MICROSCOPY

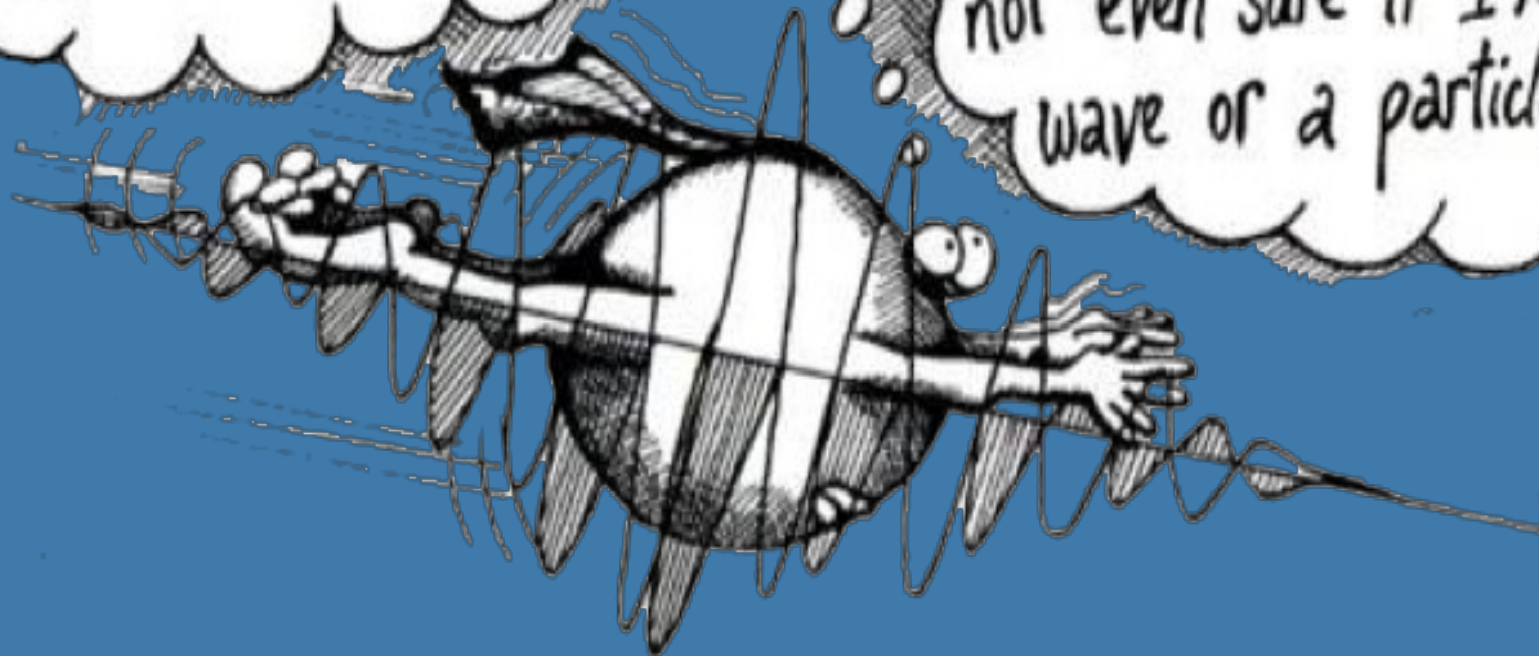
- **PRINCIPLES OF FLUORESCENCE**
 - **FLUOROPHORES / DYES**
- **THE FLUORESCENCE MICROSCOPE**
 - **OPTICAL FILTERS**





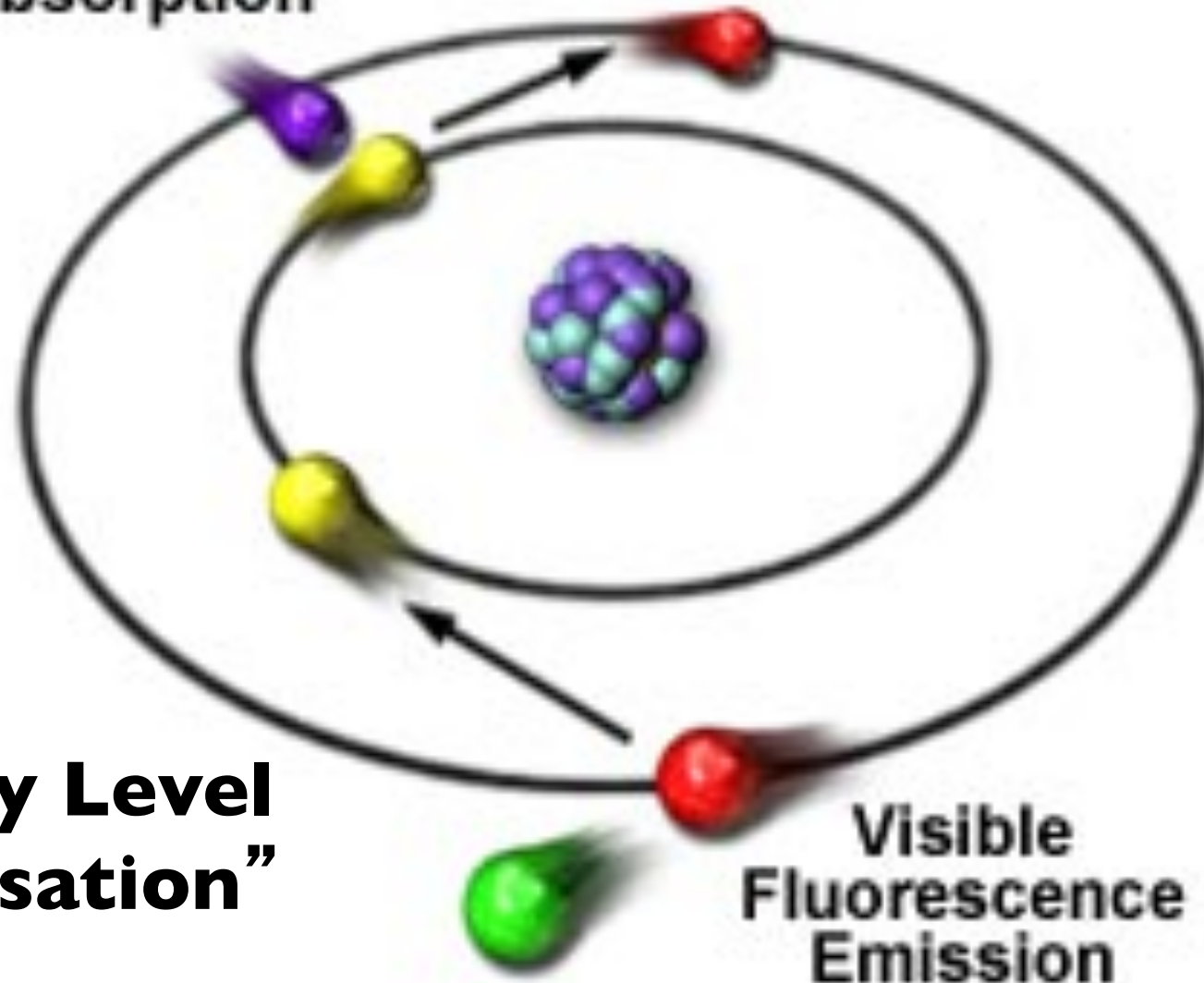
Where am I...?
Or what is my momentum...?
Or where am I...?

Oh hell...! Why worry about
all that again...? I'm
not even sure if I'm a
wave or a particle!



Stokes' Observation

UV
Absorption



**“Energy Level
Quantisation”**

Visible
Fluorescence
Emission



Processes leading to fluorescence in fluorescent dyes

1. Absorption of a photon by a Dye Molecule



2. Dye Molecule relaxes

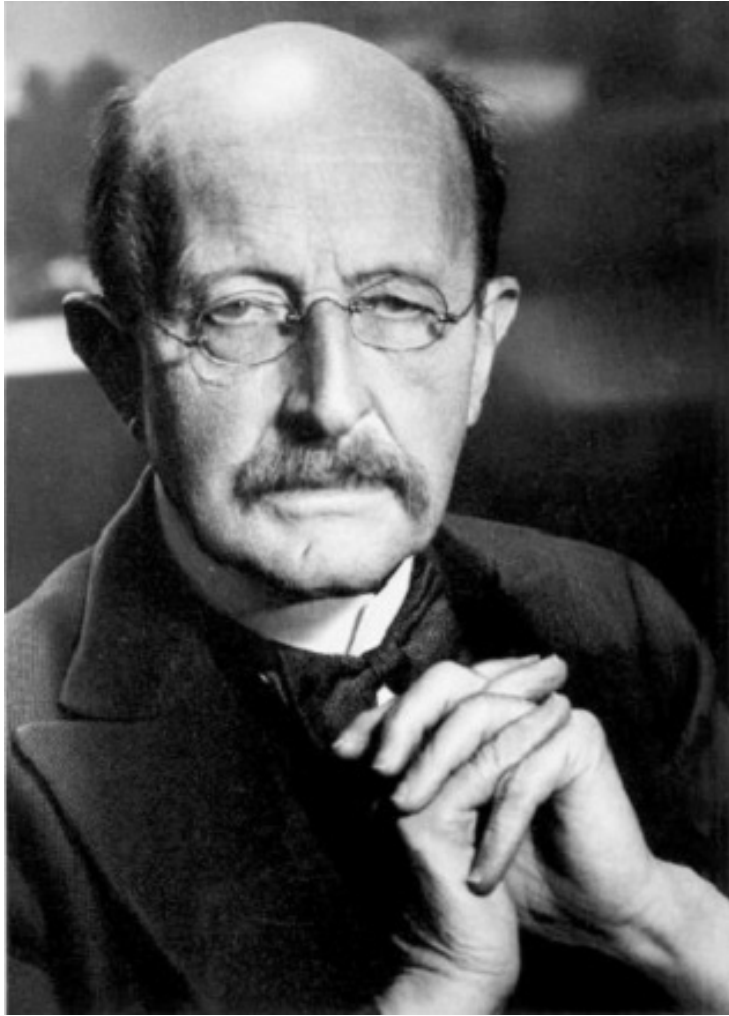


3. Dye Molecule emits a photon









Max Planck, 1858 - 1947



<http://en.wikipedia.org/>

Longer wavelength = lower energy

$$E = h\nu$$

or

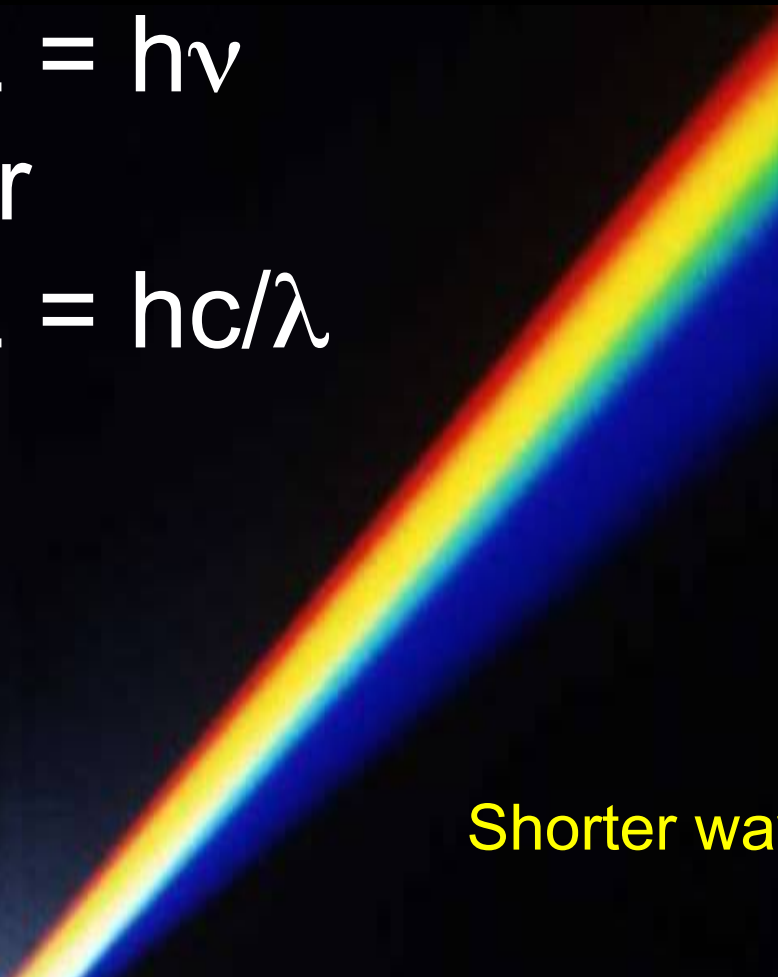
$$E = hc/\lambda$$

} Infrared

} Visible

} Ultraviolet

Shorter wavelength = higher energy

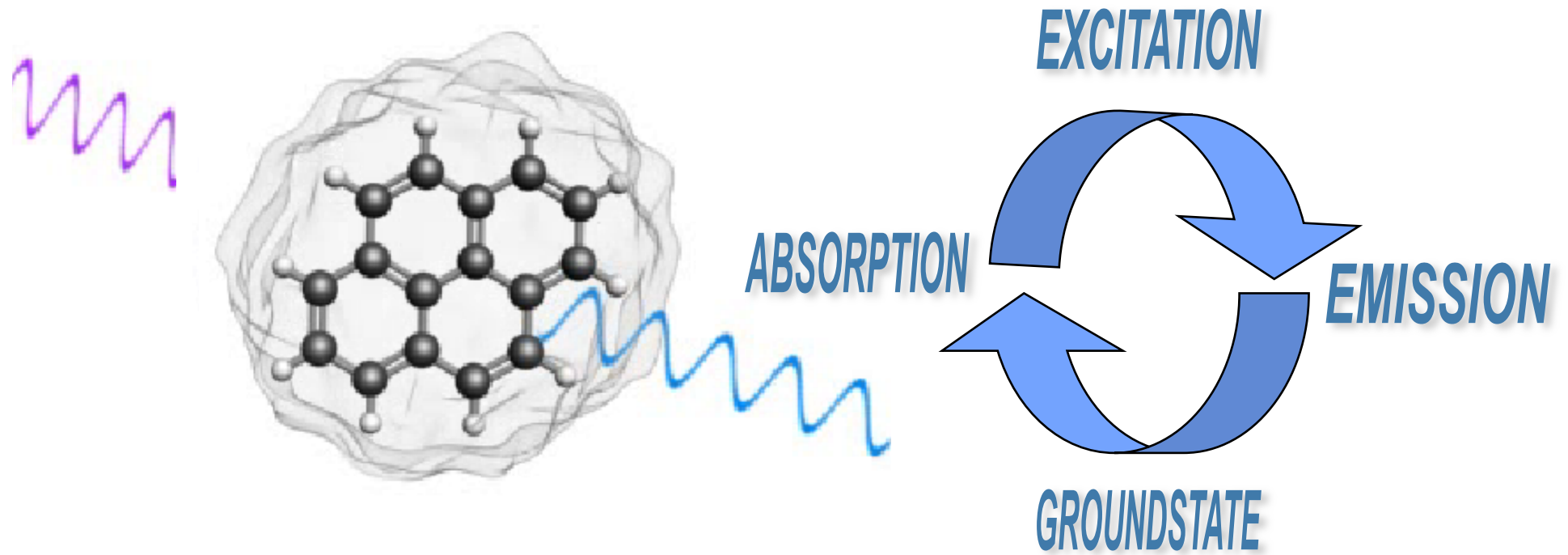


ABSORPTION OF LIGHT

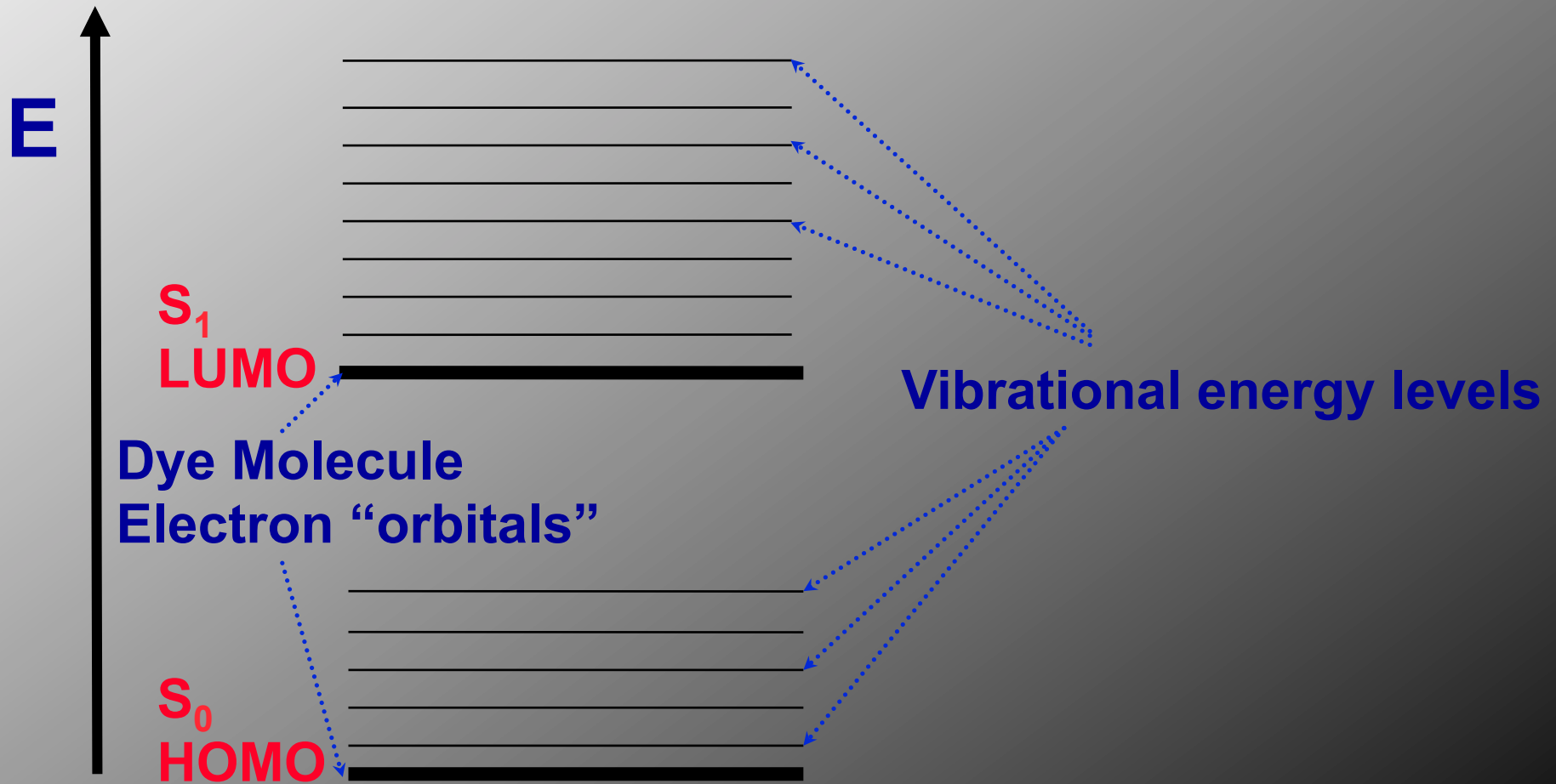
All molecules absorb light
different molecules - different wavelengths

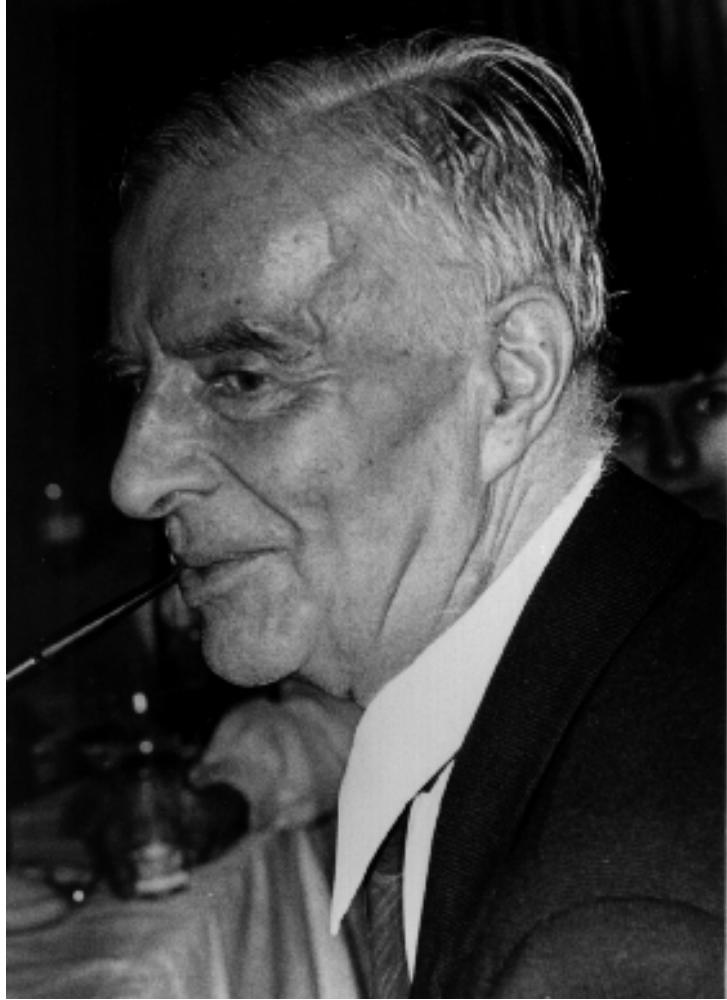
- Absorption of microwaves causes molecular rotations,
- Absorption in the infra red causes molecular bond vibrations,
- Absorption of UV/visible light causes electrons to jump to higher energy “orbitals”.





Energy Level Diagram



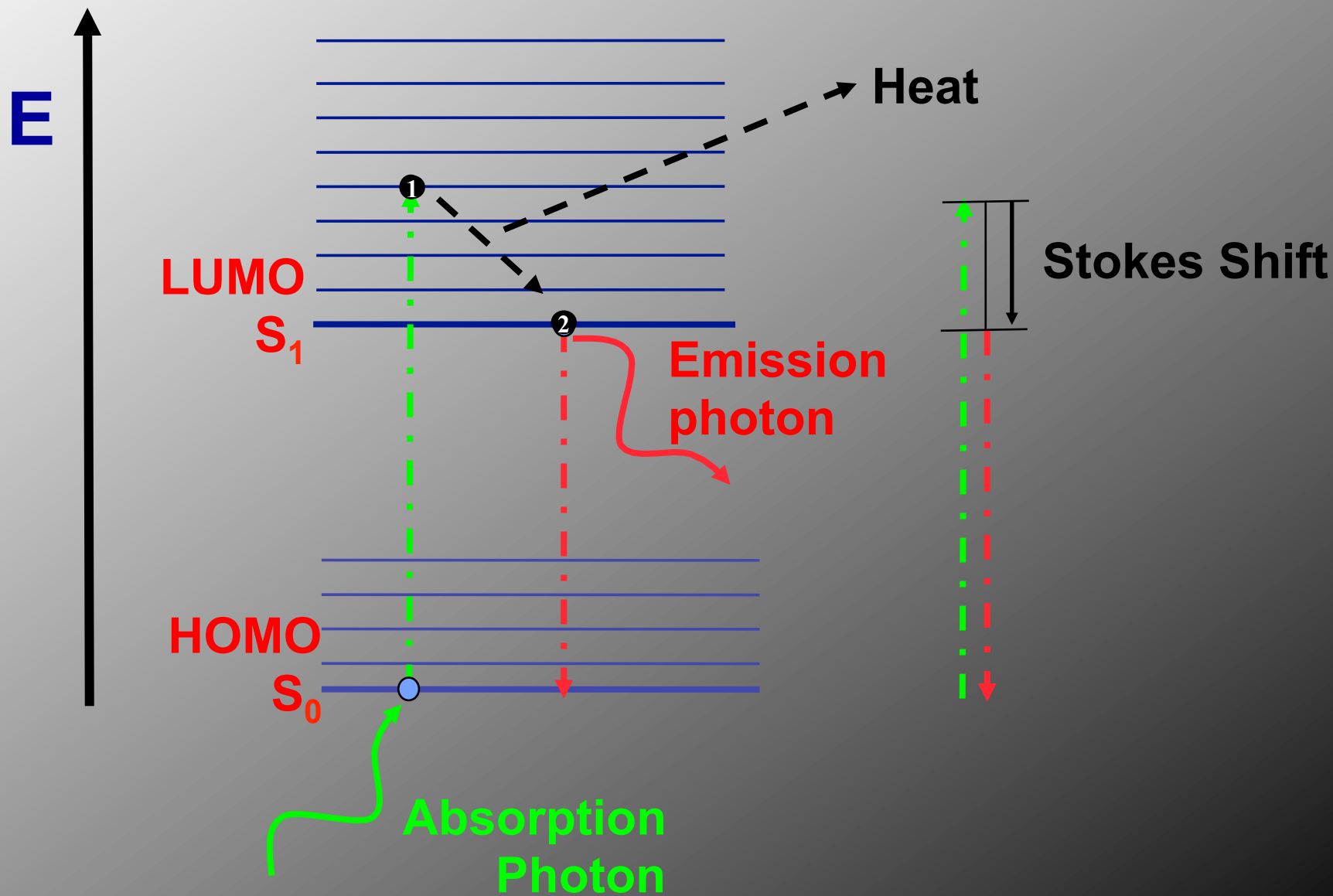


Aleksander Jabłoński, 1898 - 1980



<http://en.wikipedia.org/>

Fluorescence Absorption / Emission



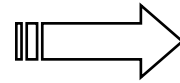


George Stokes, 1819 – 1903



<http://en.wikipedia.org/>

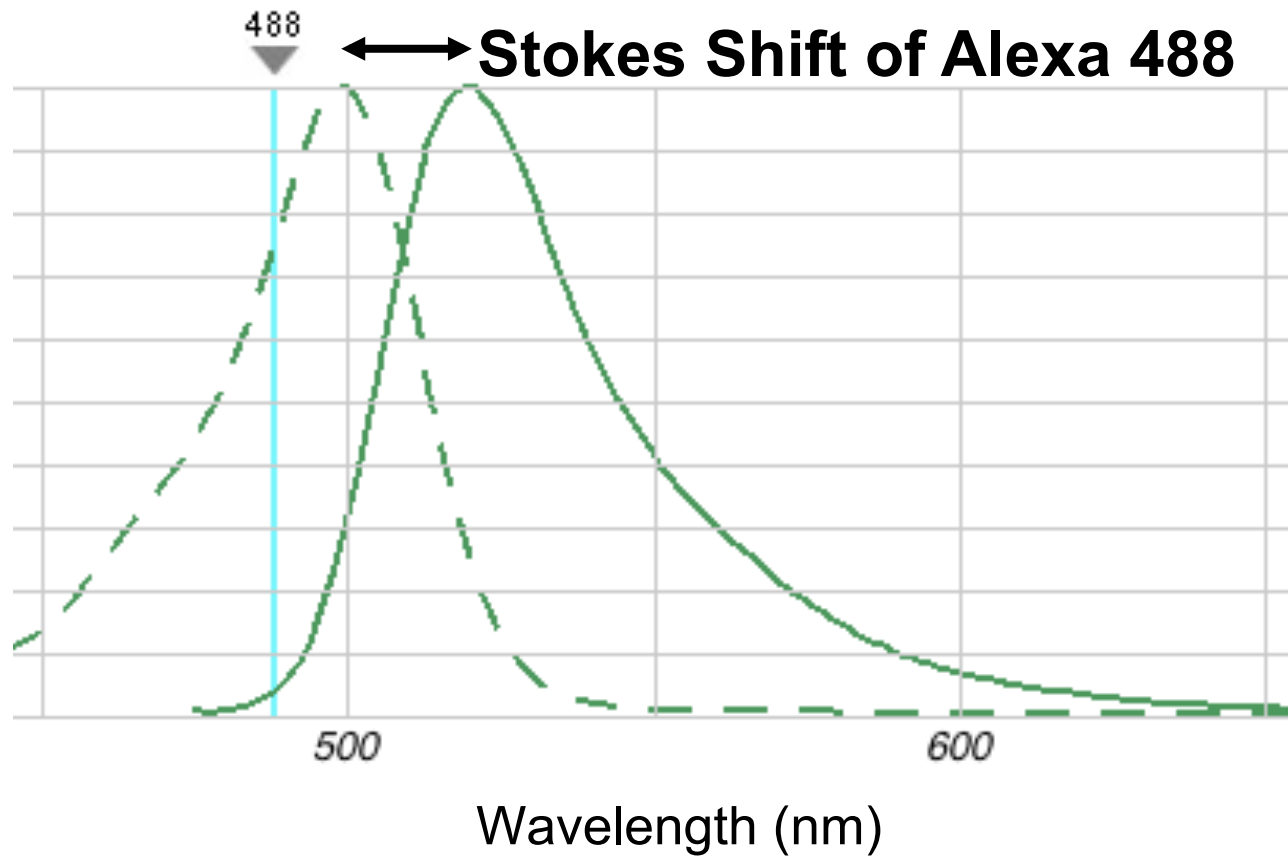
Emission has lower energy



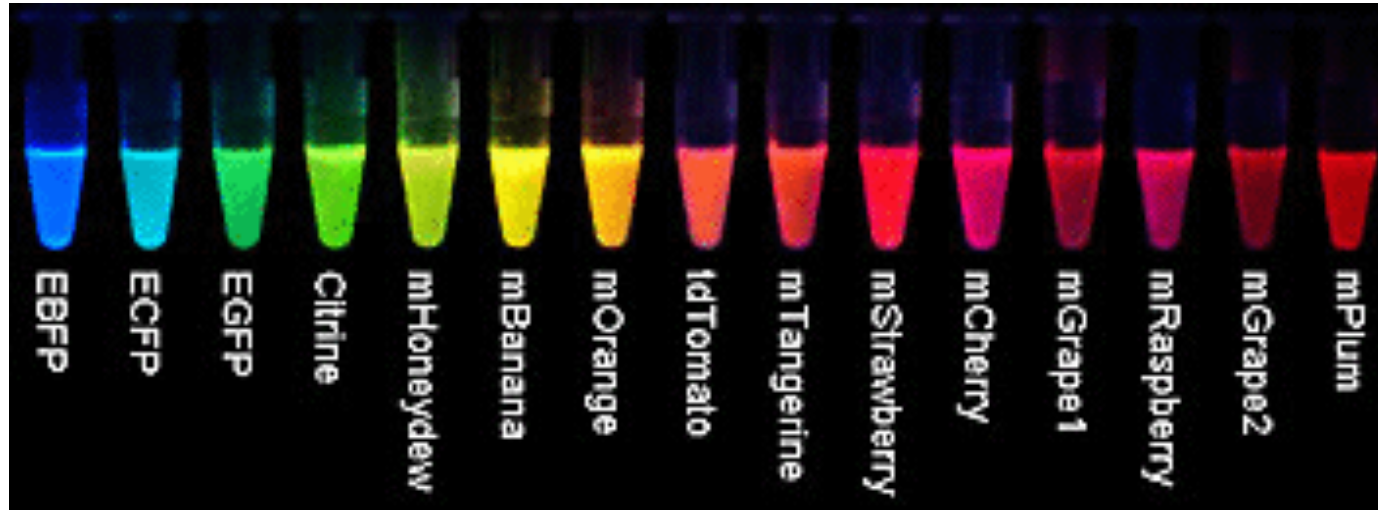
Longer wavelength

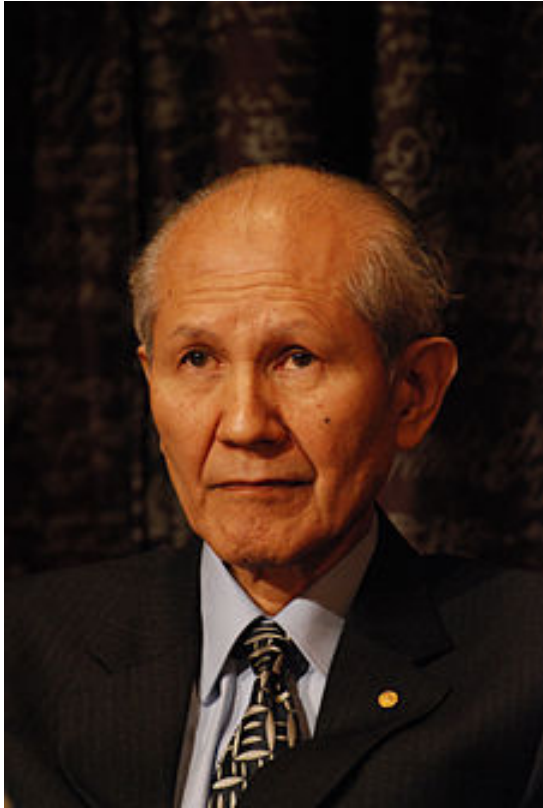
Absorption = Excitation

Emission = Fluorescence



Know your fluorophores !

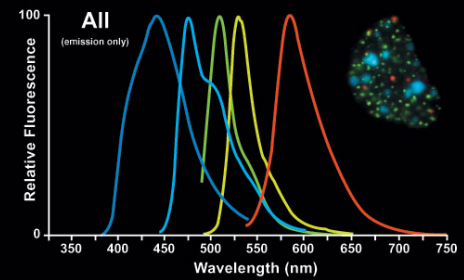
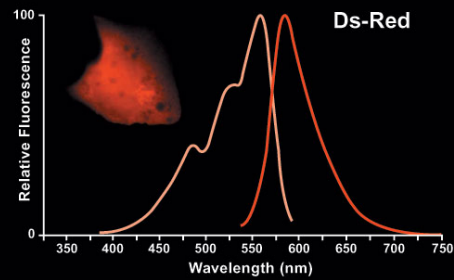
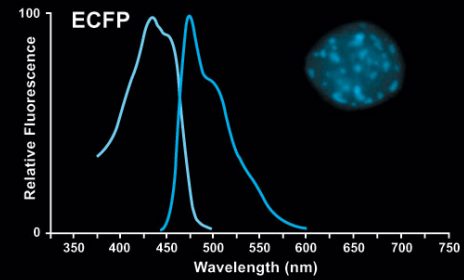
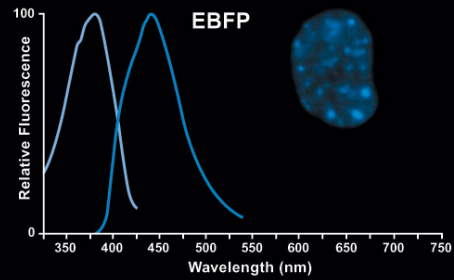
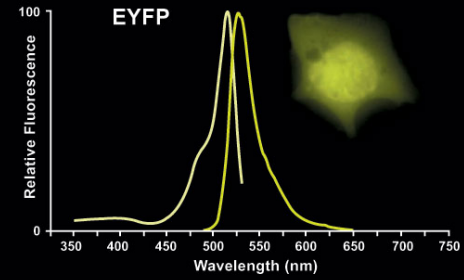
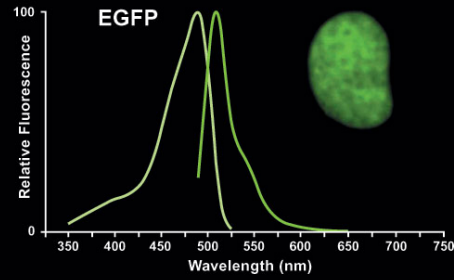




Osamu Shimomura, 1928 Martin Chalfy, 1947, Roger Tsien, 1952



<http://en.wikipedia.org/>



Fluorescent Protein Spectra

George Patterson, Rich N. Day and David Piston

Quantum Yield (QY)

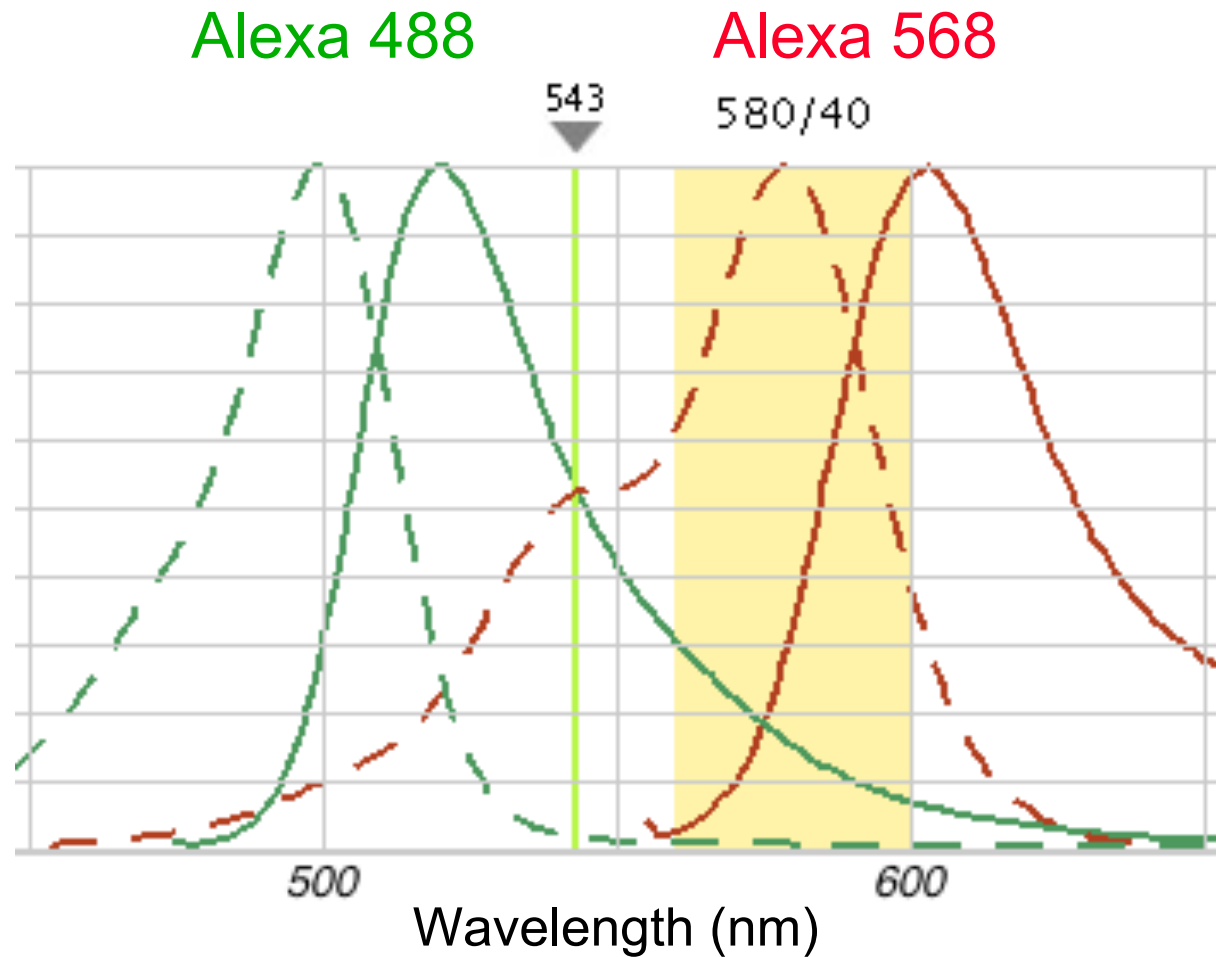
$$\text{QY} = \frac{\text{Number of photons emitted}}{\text{Number of photons absorbed}}$$



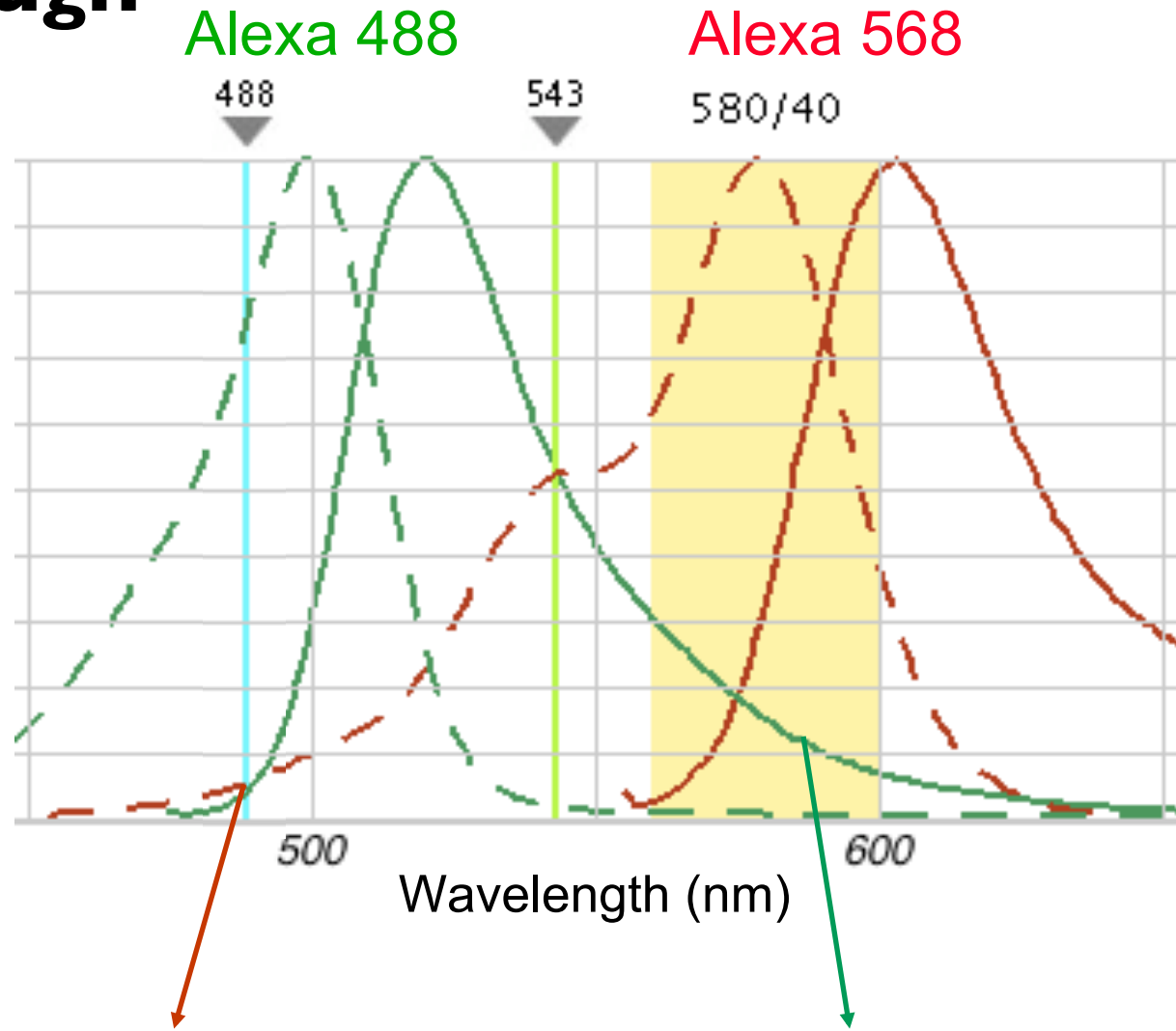
*1st TAKE HOME MESSAGE.....
KNOW YOUR
FLUOROPHORE!!!!*



Multiple colour/dye imaging...



Beware ! Crosstalk and Bleed Through



Cross talk (wrong excitation)

Bleed through (wrong emission)

PHOTOBLEACHING

